

## Synthesis of Spiro[pyrido[3,2-*b*][1,4]oxazin-2,2'-pyrans] based upon Methyl *D-arabino*-2-Hexulopyranosonate

Jens Andersch, Dieter Sicker,\* and Horst Wilde

Institut für Organische Chemie, Universität Leipzig, Talstr. 35, D-04103 Leipzig, Germany

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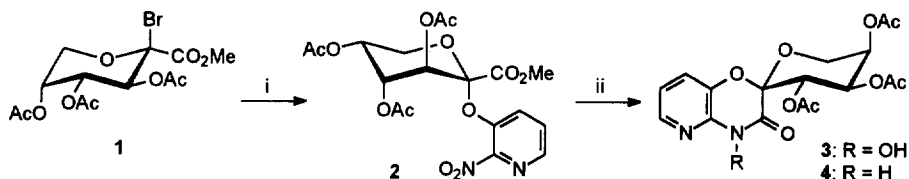
Dedicated to Professor Klaus Burger on the occasion of his 60th birthday

**Abstract:** The novel glycosyl donor **1**, derived from methyl *D-arabino*-2-hexulopyranosonate, was transformed into glycoside **2**, diastereoselectively. Catalytic hydrogenation of **2** and spontaneous reductive cyclization gave access to the spiro[pyrido[3,2-*b*][1,4]oxazin-2,2'-pyrans] **3** and **4**.  
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*D-arabino*-2-Hexulosonic acid<sup>1</sup> is an ignored carbohydrate building block though it is used on industrial scale for the production of *D*-isoascorbic acid.<sup>2</sup> We wish to report on the synthesis of a novel combination of the *D-arabino*-2-hexulonosonate unit with the pharmacological<sup>3</sup> and herbicidal active<sup>4,5</sup> 2*H*-pyrido[3,2-*b*][1,4]oxazin-3(4*H*)-one skeleton. This heterocycle-monosaccharide combination resembles some benzoxazinoid acetal glucosides naturally occurring in *Gramineae* species.<sup>6</sup>

Glycosyl donor **1**<sup>7</sup> was obtained on bromination of methyl 2,3,4,5-tetra-*O*-acetyl-β-*D-arabino*-2-hexulopyranosonate<sup>8</sup> with 33% hydrogen bromide in glacial acetic for 2 h at 20°C.



Conditions: (i) 3-hydroxy-2-nitropyridine, K<sub>2</sub>CO<sub>3</sub>, acetone, reflux; (ii) H<sub>2</sub>/Pt-C, MeOH, 20°C.

Scheme 1. Synthesis of spiro acetals **3** and **4**

Nucleophilic substitution of **1** with 3-hydroxy-2-nitropyridine gave neighbouring group assisted diastereoselectively the 2,3-*trans* glycoside **2**.<sup>9</sup> Catalytic hydrogenation of **2** and spontaneous cyclization of the hydroxylamine intermediate led to the hydroxamic acid **3** accompanied by lactam **4**.<sup>10</sup> Standard deprotection of **3** and **4** with NaOMe/MeOH gave rise to the free heterocyclic spiroacetals.<sup>11</sup>

By means of  $^1\text{H}$  NMR, the conformational change from the  ${}^2\text{C}_5$  geometry of **1** ( $J_{3,4}=10.0$ ,  $J_{4,5}=3.2$  Hz) into  ${}^5\text{C}_2$  geometry in **2** ( $J_{3,4}=4.8$ ,  $J_{4,5}=3.6$  Hz) and back to  ${}^2\text{C}_5$  in **3**, **4** ( $J_{3,4}=10.4$ - $10.6$ ,  $J_{4,5}=3.7$  Hz) was proven.

In summary, the procedure described here makes use of *D-arabino*-2-hexulopyranosonates as carbohydrate building block for a novel heterocycle-saccharide combination.

#### ACKNOWLEDGEMENTS

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- 1** (98%) of colourless powder from  $\text{CHCl}_3$ , mp. 78-80°C;  $[\alpha]_D^{21}$  -175° ( $\text{CHCl}_3$ , c 1);  $^1\text{H}$  NMR (200 MHz,  $\text{CDCl}_3$ ),  $\delta$  2.00 (s, 3H,  $\text{CH}_3$ ), 2.10 (s, 3H,  $\text{CH}_3$ ), 2.15, (s, 3H,  $\text{CH}_3$ ), 3.85 (s, 3H,  $\text{OCH}_3$ ), 4.25 (d, 1H,  $J_{6a,b}=13.4$ , H-6a), 4.33 (d, 1H,  $J_{6a,b}=13.4$  Hz, H-6b), 5.13 (dd, 1H,  $J_{3,4}=10.0$ ,  $J_{4,5}=3.2$  Hz, H-4), 5.39 (m, 1H, H-5), 5.48 (d, 1H,  $J_{3,4}=10.0$  Hz, H-3). See ref.<sup>11</sup> for  $^{13}\text{C}$  NMR and EIMS.
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- Chromatography (toluene/EtOAc 3:1 v:v) on flash Silicagel 60 (Merck) yielded **2** (82%) as yellow crystals mp. 47-49°C;  $[\alpha]_D^{28}$  +25° ( $\text{CHCl}_3$ , c 1);  $^1\text{H}$  NMR (400 MHz,  $\text{CDCl}_3$ ),  $\delta$  1.99 (s, 3H,  $\text{CH}_3$ ), 2.06 (s, 3H,  $\text{CH}_3$ ), 2.12 (s, 3H,  $\text{CH}_3$ ), 3.64 (s, 3H,  $\text{OCH}_3$ ), 3.97 (dd, 1H,  $J_{6a,6b}=10.2$ ,  $J_{5,6}=8.8$  Hz, H-6a), 4.06 (dd, 1H,  $J_{5,6}=4.4$  Hz, H-6b), 5.29 (m, 1H, H-5), 5.33 (dd, 1H,  $J_{3,4}=4.8$ ,  $J_{4,5}=3.6$  Hz, H-4), 5.43 (d, 1H, H-3), 7.44 (dd, 1H,  $J_{5,6}=4.4$ ,  $J_{4,6}=1.1$  Hz, H-5), 7.83 (dd, 1H,  $J_{4,5}=8.4$ ,  $J_{4,6}=1.1$  Hz, H-4), 8.14 (dd, 1H, H-6'). See ref.<sup>11</sup> for other analytical data.
- Chromatography on flash Silicagel 60 (Merck) with toluene/EtOAc (1:1, v:v) yielded first **4** (21%), mp. 190-191°C;  $[\alpha]_D^{25}$  +24° ( $\text{CHCl}_3$ , c 1); CD:  $\Delta\epsilon_{231}$  +19.1,  $\Delta\epsilon_{252}$  -2.7,  $\Delta\epsilon_{268}$  +0.2,  $\Delta\epsilon_{294}$  +10.6 ( $\text{CHCl}_3$ , c 0.30);  $^1\text{H}$  NMR (200 MHz,  $\text{CDCl}_3$ ),  $\delta$  2.01 (s, 3H,  $\text{CH}_3$ ), 2.05 (s, 3H,  $\text{CH}_3$ ), 2.19 (s, 3H,  $\text{CH}_3$ ), 4.03 (m, 2H, H-6'a,b), 5.42 (m, 1H, H-5'), 5.72 (d, 1H,  $J_{3,4}=10.6$  Hz, H-3'), 6.11 (dd, 1H,  $J_{3,4}=10.6$ ,  $J_{4,5}=3.7$  Hz, H-4'), 7.05 (dd, 1H,  $J_{6,7}=4.9$ ,  $J_{7,8}=7.5$  Hz, H-7), 7.48 (d, 1H,  $J_{7,8}=7.5$  Hz, H-8), 8.21 (d, 1H,  $J_{6,7}=4.9$  Hz, H-6), 12.13 (s, 1H, NH);  $^{13}\text{C}$  NMR (50 MHz,  $\text{CDCl}_3$ ),  $\delta$  21.1 ( $\text{CH}_3$ ), 21.3 ( $\text{CH}_3$ ), 21.4 ( $\text{CH}_3$ ), 66.2 (C-6'), 68.4 (C-5'), 69.8 (C-4'), 70.0 (C-3'), 98.3 (C-2), 120.0 (C-7), 126.0 (C-8), 137.0 (C-8a), 140.6 (C-4a), 141.9 (C-6), 161.0 (C-3), 170.1 (CO), 170.2 (CO), 170.8 (CO); EIMS, m/z (%): 394 ( $\text{M}^+$ , 33), 335 (6), 275 (3), 233 (5), 215 (11), 165 (8), 137 (24), 42 (100). See ref.<sup>11</sup> for other analytical data. On changing the eluent to  $\text{CHCl}_3/\text{MeOH}$  (7:3, v:v) **3** (57%) was obtained: mp. 125-127°C;  $[\alpha]_D^{22}$  -55° ( $\text{CHCl}_3$ , c 1.00); CD:  $\Delta\epsilon_{224}$  +13.5,  $\Delta\epsilon_{252}$  -8.7,  $\Delta\epsilon_{291}$  +16.9 ( $\text{CHCl}_3$ , c 0.61);  $^1\text{H}$  NMR (200 MHz,  $\text{CDCl}_3$ ),  $\delta$  1.98 (s, 3H,  $\text{CH}_3$ ), 2.01 (s, 3H,  $\text{CH}_3$ ), 2.14 (s, 3H,  $\text{CH}_3$ ), 3.79 (m, 2H, H-6'a,b), 5.36 (m, 1H, H-5'), 5.64 (d, 1H,  $J_{3,4}=10.4$  Hz, H-3'), 6.03 (dd, 1H,  $J_{3,4}=10.4$ ,  $J_{4,5}=3.7$  Hz, H-4'), 7.06 (dd, 1H,  $J_{6,7}=5.1$ ,  $J_{7,8}=7.6$  Hz, H-7), 7.48 (d, 1H,  $J_{7,8}=7.6$  Hz, H-8), 8.07 (d, 1H,  $J_{6,7}=5.1$  Hz, H-6);  $^{13}\text{C}$  NMR (50 MHz,  $\text{CDCl}_3$ ),  $\delta$  21.0 ( $\text{CH}_3$ ), 21.1 (2 x  $\text{CH}_3$ ), 66.0 (C-6'), 68.0 (C-5'), 69.8 (C-4'), 69.8 (C-3'), 100.2 (C-2), 120.7 (C-7), 126.7 (C-8), 137.6 (C-8a), 140.9 (C-4a), 141.3 (C-6), 157.5 (C-3), 170.2 (CO), 170.3 (CO), 170.8 (CO); EIMS, m/z (%): 410 ( $\text{M}^+$ , 27), 394 (3), 368 (5), 340 (2), 233 (5), 181 (5), 137 (11), 126 (10), 108 (8), 85 (9), 42 (100).
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